



Information about the subject

Degree: Bachelor of Science Degree in Biotechnology

Faculty: Faculty of Veterinary Medicine and Experimental Sciences

Code: 1101108 **Name:** General Chemistry

Credits: 6,00 **ECTS Year:** 1 **Semester:** 1

Module: Chemistry for Biomolecular Sciences

Subject Matter: Chemistry **Type:** Basic Formation

Field of knowledge: Sciences

Department: -

Type of learning: Classroom-based learning

Languages in which it is taught: Spanish

Lecturer/-s:



Module organization

Chemistry for Biomolecular Sciences

Subject Matter	ECTS	Subject	ECTS	Year/semester
Chemistry	12,00	General Chemistry	6,00	1/1
		Organic Chemistry	6,00	1/2
Chemistry Physics	6,00	Thermodynamics and Kinetics	6,00	2/2

Recommended knowledge

High School Chemistry



Learning outcomes

At the end of the course, the student must be able to prove that he/she has acquired the following learning outcomes:

- R1 The student has understood and assimilated the contents of the subject.
- R2 The student is able to solve problems or case studies related to the subject contents, by using different resources (bibliographic, IT, etc.)
- R3 The student is able to work in a laboratory, carrying out basic operations correctly and taking into account the corresponding safety standards. He/she understands the planning, development and purpose of the experience, and is able to contrast and validate the obtained results.
- R4 The student is able to write an intelligible and organized text on different aspects of the subject.
- R5 The student is able to present and defend his/her work adequately.
- R6 The student seeks bibliographic information from different sources and can analyze it with a critical and constructive spirit.
- R7 The student collaborates with the teacher and his/her peers throughout the learning process; he/she works in a team; treats everyone with respects, is proactive and fulfills the organization rules of the course.



Competencies

Depending on the learning outcomes, the competencies to which the subject contributes are (please score from 1 to 4, being 4 the highest score):

BASIC		Weighting			
		1	2	3	4
CB1	Students acquire and understand knowledge in their field of study based on general secondary education but usually reaching a level that, although supported on advanced text books, also includes aspects involving state-of-the-art knowledge specific to their area.				X
CB2	Students are able to apply knowledge to their work in a professional way and have the competences enabling them to state and defend views and opinions as well as perform problem-solving tasks in their field of study.			X	
CB3	Students are able to collect and interpret relevant data (generally in their field of study) and give opinions that involve reflection on relevant social, scientific or ethical issues.			X	
CB4	Students can communicate information, ideas, problems and solutions to a specialized or non-specialized audience.				X
CB5	Students develop the necessary learning skills to undertake further studies with a high level of autonomy.				X
GENERAL		Weighting			
		1	2	3	4
CG01	Capacity to analyze and synthesize.				X
SPECIFIC		Weighting			
		1	2	3	4
CE22	Knowing and understanding contents, principles and theories related to biotechnology.			X	



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TRANSVERSAL		Weighting			
		1	2	3	4
CT02	Capacity to organize and plan.			X	
CT03	Mastering Spanish oral and written communication.				X
CT05	Knowing and applying Basic ITC skills related to Biotechnology.				X
CT06	Capacity to manage information (capacity to look for and analyze information coming from different types of sources).			X	
CT07	Problem solving.				X
CT08	Decision making				X
CT09	Capacity to work in interdisciplinary and multidisciplinary team.			X	
CT10	Interpersonal skills.			X	
CT11	Understanding multicultural and diverse environment			X	
CT12	Critical and self-critical capacity.			X	
CT13	Ethics.				X



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CT14	Capacity to learn			x
CT15	Capacity to adapt to new situations		x	
CT16	Capacity to produce new ideas (creativity)		x	
CT17	Leadership abilities	x		
CT18	Taking initiatives and enterprising spirit		x	
CT19	Capacity to apply theoretical knowledge		x	
CT20	Research skills	x		
CT21	Sensitivity to environmental issues		x	



Assessment system for the acquisition of competencies and grading system

Assessed learning outcomes	Granted percentage	Assessment method
R1, R2, R4, R5	60,00%	Written test
R1, R2, R4, R5, R6, R7	10,00%	Submission of papers
R1, R3	30,00%	Laboratory test

Observations

In the laboratory test and the written test, a minimum mark of 5.0 out of 10 (in each) is required to pass the course. In addition, in the case of the written test, a minimum mark of 5.0 out of 10 is required for the theory part and 5.0 out of 10 for the problem part in order to be able to average between both marks and with the rest. average between the two marks and with the rest. The percentage of the theory test is divided into Theory 25%; Problems 35%. The work submission consists of a Collaborative Work.

The "laboratory test" consists of a laboratory practical. In order to be allowed to take the test, it will be compulsory to wear a lab coat, as well as the appropriate clothing according to the safety rules of the laboratories (closed shoes, long trousers, hair tied up). Failure to comply with the aforementioned rules will make it impossible for the student to take the test. Attendance at the laboratory practicals is compulsory to pass the course. Only one absence from practical sessions will be accepted, provided that it is duly justified. The teacher will keep track of attendance. Unjustified absence from a practical will result in a 10% penalty in the practical exam mark. Unjustified absence from all practical sessions will result in a penalty of 50% of the practical exam mark. Absence from more than one practice, without reaching the totality, will be studied individually to choose the penalty at the discretion of the teacher.

MENTION OF DISTINCTION:

According to Article 22 of the Regulations governing the Evaluation and Qualification of UCV Courses, the mention of "Distinction of Honor" may be awarded by the professor responsible for the course to students who have obtained, at least, the qualification of 9 over 10 ("Sobresaliente"). The number of "Distinction of Honor" mentions that may be awarded may not exceed five percent of the number of students included in the same official record, unless this number is lower than 20, in which case only one "Distinction of Honor" may be awarded.



Learning activities

The following methodologies will be used so that the students can achieve the learning outcomes of the subject:

- M1 Teacher presentation of contents, analysis of competences, explanation and in-class display of skills, abilities and knowledge.
- M2 Group work sessions supervised by the professor. Case studies, diagnostic tests, problems, field work, computer room, visits, data search, libraries, on-line, Internet, etc. Meaningful construction of knowledge through interaction and student activity.
- M3 Activities carried out in spaces with specialized equipment.
- M4 Supervised monographic sessions with shared participation..
- M5 Application of multidisciplinary knowledge.
- M6 Personalized and small group attention. Period of instruction and/or guidance carried out by a tutor to review and discuss materials and topics presented in classes, seminars, readings, papers, etc.
- M7 Set of oral and/or written tests used in initial, formative or additive assessment of the student
- M8 Group preparation of readings, essays, problem-solving, seminars, papers, reports, etc. to be presented or submitted in theoretical , practical and/or small-group tutoring sessions. Work done on the university e-learning.
- M9 Student's study: Individual preparation of readings, essays, problem-solving, seminars, papers, reports, etc. to be presented or submitted in theoretical, practical and/or small-group tutoring sessions. Work done on the university e-learning platform.



IN-CLASS LEARNING ACTIVITIES

	LEARNING OUTCOMES	HOURS	ECTS
ON-CAMPUS CLASS M1	R1, R2, R7	37,00	1,48
PRACTICAL CLASSES M2	R1, R2, R4, R5, R6, R7	5,00	0,20
LABORATORY M3	R1, R2, R3, R7	8,00	0,32
SEMINAR M4	R1, R2, R4, R7	3,00	0,12
GROUP PRESENTATION OF ASSIGNMENTS M5	R1, R4, R5, R6, R7	3,00	0,12
TUTORIAL M6	R1, R2, R5, R6	2,00	0,08
ASSESSMENT M7	R1, R2, R3, R4, R5	2,00	0,08
TOTAL		60,00	2,40

LEARNING ACTIVITIES OF AUTONOMOUS WORK

	LEARNING OUTCOMES	HOURS	ECTS
AUTONOMOUS GROUP WORK M8	R1, R2, R3, R4, R5, R6, R7	18,00	0,72
AUTONOMOUS INDIVIDUAL WORK M9	R1, R2, R3, R4, R5, R6	72,00	2,88
TOTAL		90,00	3,60



Description of the contents

Description of the necessary contents to acquire the learning outcomes.

Theoretical contents:

Content block	Contents
Didactic unit 1: structure of the atom.	<ol style="list-style-type: none">1 Exercises stoichiometry.2 The structure of the atom models: models of THOMSOM, RUTHEFORD.3 Emission and absorption spectra4 Theory of Plank5 Model of Bohr6 Sommerfeld model7 Principles of cuantica mechanics8. SCHRÖEDINGER equation9 Atomic Orbital and electron configuration
Didactic unit 2: chemical periodicity and the periodic table	<ol style="list-style-type: none">1 Triads of BÖBEREIMER, Newlands octaves, Mendeleev and Meyer, Moseley2 Properties periodical: conductivity electric, Atomic and ionics radii ionization potential, electron affinity, electronegativity.
Didactic unit 3: the chemical bond	<ol style="list-style-type: none">1. Lewis structures.2 Link Ionic. Properties of ionic compounds3 Covalent. Properties of Covalent compounds.4 Molecular orbital theory. Diagrams of molecular orbital (DOM)
Didactic unit 4: chemical equilibrium	<ol style="list-style-type: none">1. Equilibrium constant. Principle of Le Chatelier2. Espontaneity of a process3. Reaction speed4. Balances acid base pH, dissolved tampon.5. Redox equilibrium. Batteries. Corrosion6. Solubility equilibrium. Factors affecting solubility



Didactic unit 5: practice laboratory

1. Preparation of solutions, obtaining and precipitate separation.
2. Acid-base balances and pH measurement.
3. Preparation of buffer solutions.
4. Determination of the density of liquids.
5. Comparison of the concentration of acetylsalicylic acid in an aspirin Bayer and another generic.

Organization of the practical activities:

	Content	Place	Hours
PR1.	Preparation of solutions, obtaining and separation of precipitates.	Laboratory	2,00
PR2.	Acid-base equilibrium and pH measurement.	Laboratory	2,00
PR3.	Preparations of buffer solutions	Laboratory	2,00
PR4.	Determination of the density of liquids	Laboratory	2,00
PR5.	Comparison of the concentration of acetylsalicylic acid in an aspirin Bayer and another generic.	Laboratory	2,00
PR6.	Practical sessions	Lecture room	3,00



Temporary organization of learning:

Block of content	Number of sessions	Hours
Didactic unit 1: structure of the atom.	3,00	6,00
Didactic unit 2: chemical periodicity and the periodic table	4,00	8,00
Didactic unit 3: the chemical bond	6,00	12,00
Didactic unit 4: chemical equilibrium	12,00	24,00
Didactic unit 5: practice laboratory	5,00	10,00



References

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